



King's Research Portal

DOI:

[10.3390/molecules23092253](https://doi.org/10.3390/molecules23092253)

Document Version

Publisher's PDF, also known as Version of record

[Link to publication record in King's Research Portal](#)

Citation for published version (APA):

Rundstadler, T. K., Eskandari, A., Norman, S. M., & Suntharalingam, K. (2018). Polypyridyl zinc(II)-Indomethacin complexes with potent anti-breast cancer stem cell activity. *Molecules*, 23(9), [2253].
<https://doi.org/10.3390/molecules23092253>

Citing this paper

Please note that where the full-text provided on King's Research Portal is the Author Accepted Manuscript or Post-Print version this may differ from the final Published version. If citing, it is advised that you check and use the publisher's definitive version for pagination, volume/issue, and date of publication details. And where the final published version is provided on the Research Portal, if citing you are again advised to check the publisher's website for any subsequent corrections.

General rights

Copyright and moral rights for the publications made accessible in the Research Portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognize and abide by the legal requirements associated with these rights.

- Users may download and print one copy of any publication from the Research Portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the Research Portal

Take down policy

If you believe that this document breaches copyright please contact librarypure@kcl.ac.uk providing details, and we will remove access to the work immediately and investigate your claim.

Communication

Polypyridyl Zinc(II)-Indomethacin Complexes with Potent Anti-Breast Cancer Stem Cell Activity

Tiffany K. Rundstadler, Arvin Eskandari, Sarah M. Norman and Kogularamanan Suntharalingam * 

Department of Chemistry, King's College London, London SE1 1DB, UK; tiffany.rundstadler@etu.unistra.fr (T.K.R.); arvin.eskandari@kcl.ac.uk (A.E.); sarah.norman@kcl.ac.uk (S.M.N.)

* Correspondence: kogularamanan.suntharalingam@kcl.ac.uk; Tel.: +44-207-848-2595

Received: 17 August 2018; Accepted: 31 August 2018; Published: 4 September 2018



Abstract: Cancer stem cells (CSCs) are thought of as a clinically pertinent subpopulation of tumors, partly responsible for cancer relapse and metastasis. Research programs aimed at discovering anti-CSC agents have largely focused on biologics and purely organic molecules. Recently, we showed that a family of redox-active copper(II) complexes with phenanthroline-based ligands and nonsteroidal anti-inflammatory drugs (NSAIDs) such as indomethacin, are capable of potently and selectively killing breast CSCs. Herein we present analogous redox-inactive, zinc(II)-phenanthroline-indomethacin complexes with the ability to kill breast CSCs and bulk breast cancer cells with equal potency (in the submicro- or micromolar range). A single dose of the zinc(II) complexes could theoretically be administered to eliminate whole tumor populations. Excitingly, some of the zinc(II) complexes decrease the growth and viability of mammospheres to a comparable or higher degree than salinomycin, a compound known to effectively kill breast CSCs. As far as we are aware this is the first report to examine the anti-breast CSC activity of zinc(II)-containing compounds.

Keywords: metallopharmaceuticals; bioinorganic chemistry; zinc; nonsteroidal anti-inflammatory drug

1. Introduction

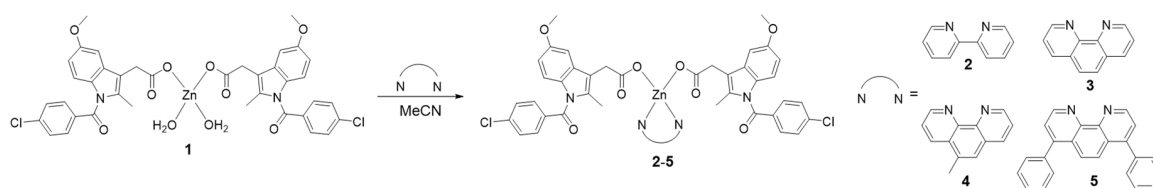
Cancer is one of the most widespread illnesses in the modern world. It is a complex multigenetic condition associated with the accumulation of irregular cells, resulting from uncontrolled cell division [1]. A subpopulation of tumors, termed cancer stem cells (CSCs) is linked to metastasis and cancer relapse, the major causes of cancer associated fatalities [2–4]. CSCs have the ability to differentiate, self-renew, and form aggressive secondary or tertiary tumors [5,6]. Current cancer therapies (chemotherapy, radiotherapy, and surgery) are able to eliminate the bulk of cancer cells but are incapable of removing CSCs [7–11]. Therefore, new generations of cancer treatments must have the capability of removing entire tumor populations, including CSCs, to have durable clinical outcomes. There are many academic- and pharmaceutical-led research programs underway with the aim of developing clinically viable anti-CSC agents, however, most of them have focused on biologics or purely organic compounds [12]. We and others have recently shown that the diversity and versatility of metals can be harnessed to develop inorganic compounds with promising anti-CSC activities [13].

Platinum(II) complexes like cisplatin, carboplatin, and oxaliplatin are routinely employed in the clinic to overcome various cancers, and a lot of effort has been put into shedding light on their cellular and molecular-level mechanisms of action [14–17]. However, these platinum(II) anticancer agents are unable to remove CSCs at their therapeutically administered doses [18–20]. Recent efforts have shown that chemical adducts between carrier/scaffold molecules and bioactive metabolites display promising anticancer activities, albeit their anti-CSC potencies are unexplored [21–25]. Copper(II) complexes with phenanthroline ligands and nonsteroidal anti-inflammatory drugs (NSAIDs) such

as indomethacin have been recently recognized as an effective class of anti-CSCs agents [26,27]. NSAIDs inhibit the inducible cyclooxygenase isoenzyme, COX-2, which is related to CSC renewal and proliferation [28–30]. The copper(II)-phenanthroline-indomethacin complexes kill breast CSCs in the micro- and sub-micromolar range, and show up to 3-fold lower potency towards bulk breast cancer cells [26,27]. Cell-based studies identified that the copper(II)-phenanthroline-indomethacin complexes prompt CSC death by increasing intracellular ROS levels (presumably via Fenton-like reactions) and inhibiting COX-2 [26,27]. The effectiveness of the copper(II)-phenanthroline-indomethacin complexes is believed to be related to the susceptibility of CSCs to variations in intracellular redox levels, and the relatively high expression of COX-2 in certain CSCs. To investigate whether redox activity is an indispensable attribute for anti-CSC activity (within this class of metal-phenanthroline-NSAID complexes), we sought to develop structurally analogous zinc(II) complexes and determine their CSC potency, given that zinc(II) is redox-inactive under physiological conditions. Herein, we present the synthesis, characterization, stability studies, and breast CSC toxicity of a family of redox-silent zinc(II)-phenanthroline-indomethacin complexes.

2. Results and Discussion

The zinc(II) complexes, **2–5** were synthesized as outlined in Scheme 1. Diaquabis(η^2 -O,O'-indomethacin)zinc(II), **1** [31] was reacted with equimolar amounts of the appropriate polypyridyl ligand (2,2'-bipyridine for **2**, 1,10-phenanthroline for **3**, 5-methyl-1,10-phenanthroline for **4**, or 4,7-diphenyl-1,10-phenanthroline for **5**) in acetonitrile. The zinc(II) complexes, **2–5** were obtained in good yields (70–85%) as yellow or orange solids and were fully characterized by ^1H NMR, infrared spectroscopy, and elemental analysis (Figures S1–S6 in the Supplementary Materials). The coordination of two indomethacin moieties and one polypyridyl ligand to zinc in **2–5** was confirmed by the 2:1, indomethacin:polypyridyl ligand integration ratio in the ^1H NMR spectra of **2–5** (Figures S1–S4). Furthermore, the formation of **2–5** was evidenced by the shift of the aromatic proton signals relative to **1** (Figure S5). According to the IR spectra of **2–5**, $\nu_{\text{asym}}(\text{CO}_2)$ and $\nu_{\text{sym}}(\text{CO}_2)$ stretching bands were observed at $1589\text{--}1594\text{ cm}^{-1}$ and $1359\text{--}1373\text{ cm}^{-1}$ (Figure S6). The variation in the $\nu_{\text{asym}}(\text{CO}_2)$ and $\nu_{\text{sym}}(\text{CO}_2)$ stretching bands for **2–5** was $221\text{--}234\text{ cm}^{-1}$, indicative of a monodentate binding mode for the carboxylate group on indomethacin to the zinc center (as depicted in Scheme 1) [32,33]. This is comparable to the $\nu_{\text{asym}}(\text{CO}_2)$ and $\nu_{\text{sym}}(\text{CO}_2)$ stretching band differences reported for related zinc(II) complexes containing indomethacin and polypyridyl ligands [34,35]. The high chemical purity of **2–5** was shown by elemental analysis studies.



Scheme 1. Representation of the synthetic route used to prepare the zinc(II)-phenanthroline complexes with two indomethacin units, **2–5**, from diaquabis(η^2 -O,O'-indomethacin)zinc(II), **1** in acetonitrile.

The shake-flask method was used to determine the lipophilicity ($\text{Log}P$) of **2–5**. This involves determining the amount of a given compound partitioned between octanol and water using UV-Vis spectroscopy. The $\text{Log}P$ values for **2–5** varied from 0.89 to 1.67 (Table S1). The size of the polypyridyl ligand has a clear effect on lipophilicity, with $\text{Log}P$ values increasing according to the following order; **2** < **3** < **4** < **5**. The hydrophobicity of **2–5** implies that the zinc(II) complexes are likely to easily enter cells. In order to determine the stability of **2** (used as a demonstrative compound of the zinc(II) complexes) in biologically relevant solutions, UV-Vis spectroscopy studies were performed. The metal-perturbed $\pi\text{--}\pi^*$ and metal-to-ligand charge-transfer (MLCT) absorption bands of **2** ($50\text{ }\mu\text{M}$) in PBS:DMSO (200:1)

were constant for 24 h at 37 °C (Figure S7), indicative of stability. The same UV-Vis bands associated to **2** (50 µM) in mammary epithelial cell growth medium (MEGM)/DMSO (200:1) remained unaltered over the course of 24 h at 37 °C, suggestive of stability in conditions required for cellular studies (Figure S8).

The antiproliferative effect of **2–5** against bulk breast cancer cells (HMLER) and breast CSC-like cells (HMLER-shEcad) was studied using the MTT assay. Dose–response curves (Figure 1A,B) were used to calculate the IC₅₀ values, which are shown in Table 1. In general, the zinc(II) complexes, **2–5** exhibited equipotency against HMLER and HMLER-shEcad cells, in the micro- or sub-micromolar range. Therefore, **2–5** have the potential ability to eliminate whole tumor cell populations with a single micro- or sub-micromolar dose. Clinically approved platinum(II) agents, cisplatin and carboplatin displayed great toxicity towards CSC-depleted HMLER cells than CSC-enriched HMLER-shEcad cells, suggestive of non-CSC (bulk cancer cell) selectivity (Table 1) [27]. This is consistent with the tendency of platinum(II) anticancer agents to encourage CSC enrichment rather than CSC depletion [18]. Notably, the potency of **2–5** against CSC-like HMLER-shEcad cells was up to 6-fold greater than salinomycin, an anti-breast CSC compound. Previous work has shown that indomethacin is non-toxic towards bulk cancer cells and CSC-like cells (Table 1) [26]. The current results indicate that the potency of indomethacin against breast CSC-like cells and bulk breast cancer cells is improved by chelating it to a zinc(II)-polypyridyl core. The CSC potency of **2–5** is similar to the most effective copper(II)-phenanthroline-indomethacin complexes previously identified by us [26,27]. This shows that CSC-potent metal-phenanthroline-NSAID complexes can be developed with redox-inactive metals such as zinc(II), and implies that redox activity is not essential for anti-CSC activity within this class of compounds.

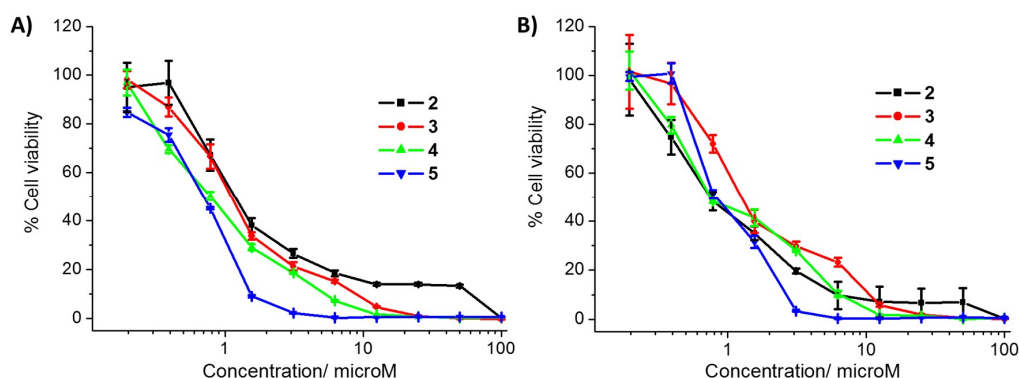


Figure 1. Compound concentration versus percentage cell viability plots for **2–5** against (A) bulk breast cancer cells (HMLER) and (B) breast CSC-like cells (HMLER-shEcad cells) (72 h incubation). The difference in potency of **2–5** towards HMLER-shEcad cells and HMLER cells is not significant (Student *t*-test, *p* > 0.05).

To measure therapeutic potential, the potency of **2–5** towards human embryonic kidney (HEK 293T) cells was investigated. The potency of **2–5** towards HEK 293T cells is highly dependent on the polypyridyl ligand and increases in the following order; **2** < **3** < **4** < **5** (Table 1 and Figure S9). Promisingly the 2,2'-bipyridine- and 1,10-phenanthroline-bearing complexes, **2** and **3** were significantly less potent toward HEK 293T cells (*p* < 0.05, up to 17-fold for **2**) than HMLER and HMLER-shEcad cells, demonstrating preferred toxicity towards breast cancer cells (bulk and CSC-like cells) over healthy cells.

To further probe the anti-CSC potential of the zinc(II) complexes, **2–5**, the mammosphere assay was performed. This assay assesses the aptitude of chemical agents to stop three-dimensional spheroid formation from breast CSCs in suspension [36]. The method also provides a reliable gauge of in vivo potential, as spheroids are closer in structure to solid tumors than monolayer systems.

Treatment of HMLER-shEcad cells in suspension, with 2–5 (at their respective IC_{20} value for 5 days) noticeably decreased the number and size of mammospheres formed (Figure 2A,B). Treatment with salinomycin under identical conditions induced a similar effect (Figure 2A,B). Notably, 4 and 5 reduced mammosphere formation (43–46% decrease in mammospheres formed) to a better degree than salinomycin (37% decrease in mammospheres formed). Previous work has shown that indomethacin has a poor mammosphere inhibitory effect [14]. TOX8, a resazurin-based compound was used to determine mammosphere viability after treatment with 2–5. The zinc(II) complexes, 3–5 displayed micromolar potency towards mammospheres, with the 4,7-diphenyl-1,10-phenanthroline-bearing complex, 5 exhibiting 7-fold higher potency than salinomycin under identical conditions (Table 1 and Figure 2C) [37]. The ineffectiveness of 2 against mammospheres ($IC_{50} > 133 \mu M$) (Figure 2C) was surprising considering its high potency (sub-micromolar) against monolayer cell cultures of HMLER-shEcad cells (Table 1). The zinc(II) complexes, 3–5 displayed similar mammosphere potency to analogous copper(II)-phenanthroline-indomethacin complexes [13], indicating that redox activity may not be vital for anti-CSC activity within this class of metal-phenanthroline-NSAID complexes.

Table 1. IC_{50} values of 2–5, salinomycin, cisplatin, and carboplatin against HMLER, HMLER-shEcad, and human embryonic kidney (HEK) 293T cells, and HMLER-shEcad mammospheres. Incubation times of 72 h or 120 h were used. IC_{50} values and errors represent the mean value of three independent experiments and the associated standard deviations, respectively.

Compound	HMLER IC_{50} (μM)	HMLER-shEcad IC_{50} (μM)	HEK 293T IC_{50} (μM)	Mammosphere IC_{50} (μM)
2	1.2 ± 0.02	0.7 ± 0.04	20.0 ± 3.9	>133
3	1.1 ± 0.05	1.1 ± 0.08	2.3 ± 0.1	44.1 ± 0.6
4	0.8 ± 0.03	0.7 ± 0.07	0.9 ± 0.2	44.2 ± 1.4
5	0.7 ± 0.03	0.9 ± 0.08	0.3 ± 0.02	2.7 ± 0.03
cisplatin ¹	3.4 ± 0.5	4.9 ± 0.4	n.d.	n.d.
carboplatin ¹	67.3 ± 2.8	72.3 ± 7.9	n.d.	n.d.
salinomycin ¹	11.4 ± 0.4	4.2 ± 0.3	n.d.	18.5 ± 1.5

¹ Taken from reference [26,27,37]. n.d. not determined.

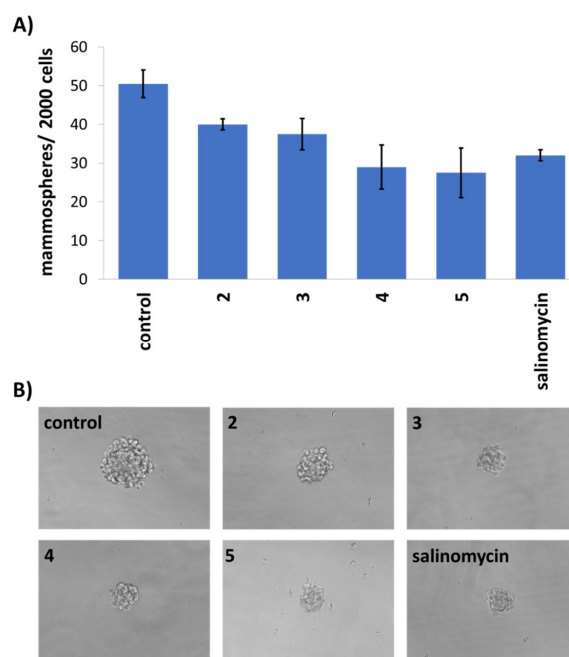


Figure 2. Cont.

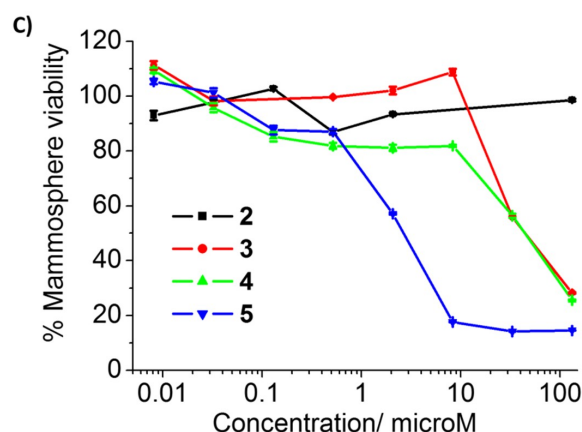


Figure 2. (A) Representation of the number of mammospheres formed from HMLER-shEcad cell suspensions treated with 2–5 or salinomycin for 5 days at their respective IC_{20} values. Standard deviation was used to calculate the associated errors. (B) Bright-field images (taken using an inverted microscope) representative of untreated HMLER-shEcad mammospheres and those treated with 2–5 or salinomycin for 5 days at their respective IC_{20} values ($\times 20$ magnification). (C) Compound concentration versus percentage mammosphere viability plots for 2–5 against HMLER-shEcad mammospheres (5 days incubation).

In summary we present a series of zinc(II)-phenanthroline-indomethacin complexes capable of killing bulk breast cancer cells and breast CSCs indiscriminately in the submicro- and micromolar range. Three of the zinc(II) complexes, 3–5 also inhibit mammosphere formation and reduce mammosphere viability to a similar or greater extent than salinomycin. Overall, the CSC potency of the redox-inactive zinc(II)-phenanthroline-indomethacin complexes is similar to the most effective redox-active copper(II)-phenanthroline-indomethacin complexes previously identified by us. This suggests that redox activity is not an essential criterion for anti-CSC activity within this class of metal-phenanthroline-NSAID complexes. Future studies will aim to elucidate the cellular mechanism of action of the zinc(II)-phenanthroline-indomethacin complexes, 2–5 in CSCs and determine their translatable scope.

3. Materials and Methods

3.1. General Procedures

No special conditions (i.e., nonatmospheric conditions) were used to carry out the synthetic methods reported in this manuscript. A BrukerAvance 400 MHz Ultrashield NMR spectrometer (Bruker, Billerica, MA, USA) was used to measure the 1H NMR spectra of 1–5. An infrared (IR) Affinity-1S Shimadzu spectrophotometer (Shimadzu Corporation, Kyoto, Japan) was used to measure the IR spectra of 2–5. Commercial services at London Metropolitan University were used to carry out the elemental analysis studies. 2,2'-bipyridine, 1,10-phenanthroline, 5-methyl-1,10-phenanthroline, 4,7-diphenyl-1,10-phenanthroline, and indomethacin were bought from Sigma Aldrich (St. Louis, MO, USA). Diaquabis(η^2 -O,O'-indomethacin)zinc(II), 1 was synthesized following a reported method [31]. For all biophysical and cellular studies, DMSO was used to prepare a 10 mM solution of 2–5. The solutions were then diluted with the appropriate biological solution to the working concentration(s).

3.2. Synthesis of $Zn(indomethacin)_2(2,2'-bipyridine)$ (2)

To a solution suspension of 1 (500.3 mg, 0.61 mmol) in acetonitrile (25 mL), 2,2'-bipyridine (96.3 mg, 0.61 mmol) was added and the resulting mixture was stirred at 60 °C for 16 h. The mixture was then cooled to room temperature and the solvent was removed to ca. 10 mL. The concentrated

solution was cooled to 4 °C to yield a precipitate, which as filtered and washed with water (20 mL) and ethanol (20 mL), to yield **2** as a yellow solid (440.6 mg, 77%); ¹H NMR (400 MHz, DMSO-d₆) δ 8.65 (d, 2H), 8.51 (d, 2H), 8.14 (t, 2H), 7.64–7.55 (m, 10H), 6.94–6.90 (m, 4H), 6.65 (dd, 2H), 3.66 (s, 6H), 3.45 (s, 4H), 2.10 (s, 6H); IR (solid state, cm^{−1}): 1677, 1654, 1639, 1594, 1560, 1522, 1478, 1459, 1433, 1395, 1373, 1358, 1324, 1290, 1256, 1219, 1181, 1148, 1091, 1069, 1035, 1012, 930, 844, 836, 753, 686, 630, 551; Anal. Calcd. For C₄₈H₃₈Cl₂N₄O₈Zn·2H₂O: C, 59.36; H, 4.36; N, 5.77. Found: C, 59.50; H, 3.25; N, 6.06.

3.3. Synthesis of Zn(indomethacin)₂(1,10-phenanthroline) (**3**)

To a solution suspension of **1** (500.3 mg, 0.61 mmol) in acetonitrile (40 mL), 1,10-phenanthroline (115.4 mg, 0.64 mmol) was added and the resulting mixture was stirred at 60 °C for 16 h. The mixture was then cooled to room temperature and the solvent was removed to ca. 10 mL. The concentrated solution was cooled to 4 °C to yield a precipitate, which as filtered and washed with water (20 mL) and ethanol (20 mL), to yield **3** as a yellow solid (423.9 mg, 72%); ¹H NMR (400 MHz, DMSO-d₆) δ 8.91 (bs, 2H), 8.75 (d, 2H), 8.13 (s, 2H), 7.88 (dd, 2H), 7.60 (s, 8H), 6.87–6.85 (m, 4H), 6.60 (dd, 2H), 3.59 (s, 6H), 3.40 (s, 4H), 2.05 (s, 6H); IR (solid state, cm^{−1}): 1672, 1623, 1593, 1559, 1525, 1480, 1457, 1431, 1404, 1371, 1359, 1325, 1292, 1258, 1231, 1220, 1182, 1148, 1145, 1092, 1077, 1039, 1013, 926, 855, 847, 787, 757, 727, 693, 670, 629, 550; Anal. Calcd. For C₅₀H₃₈Cl₂N₄O₈Zn: C, 62.61; H, 3.99; N, 5.84. Found: C, 62.47; H, 3.84; N, 5.96.

3.4. Synthesis of Zn(indomethacin)₂(5-methyl-1,10-phenanthroline) (**4**)

To a solution suspension of **1** (500.3 mg, 0.61 mmol) in acetonitrile (25 mL), 5-methyl-1,10-phenanthroline (153.6 mg, 0.79 mmol) was added and the resulting mixture was stirred at 60 °C for 16 h. The mixture was then cooled to room temperature and the solvent was removed to ca. 10 mL. The concentrated solution was cooled to 4 °C to yield a precipitate, which as filtered and washed with water (20 mL) and ethanol (20 mL), to yield **4** as a yellow solid (418.2 mg, 70%); ¹H NMR (400 MHz, DMSO-d₆) δ 8.91–8.80 (m, 3H), 8.66 (d, 1H), 7.98 (s, 1H), 7.93–7.85 (m, 2H), 7.61 (s, 8H), 6.88–6.82 (m, 4H), 6.61 (dd, 2H), 3.58 (s, 6H), 3.39 (s, 4H), 2.78 (s, 3H), 2.05 (s, 6H); IR (solid state, cm^{−1}): 1675, 1634, 1589, 1563, 1522, 1477, 1455, 1432, 1395, 1369, 1357, 1320, 1290, 1256, 1223, 1178, 1148, 1092, 1069, 1032, 1017, 994, 923, 837, 800, 759, 729, 669, 553; Anal. Calcd. For C₅₁H₄₀Cl₂N₄O₈Zn·H₂O: C, 61.80; H, 4.27; N, 5.65. Found: C, 61.43; H, 4.39; N, 5.76.

3.5. Synthesis of Zn(indomethacin)₂(4,7-diphenyl-1,10-phenanthroline) (**5**)

To a solution suspension of **1** (500.3 mg, 0.61 mmol) in acetonitrile (25 mL), 4,7-diphenyl-1,10-phenanthroline (205.9 mg, 0.62 mmol) was added and the resulting mixture was stirred at 60 °C for 16 h. The mixture was then cooled to room temperature and the solvent was removed to ca. 10 mL. The concentrated solution was cooled to 4 °C to yield a precipitate, which as filtered and washed with water (20 mL) and ethanol (20 mL), to yield **5** as a orange solid (579.9 mg, 85%); ¹H NMR (400 MHz, DMSO-d₆) δ 8.98 (bs, 2H), 8.05 (bs, 2H), 7.87 (bs, 2H), 7.63–7.58 (m, 18H), 6.94 (bs, 2H), 6.89 (d, 2H), 6.63 (d, 2H), 3.61 (s, 6H), 3.53 (bs, 4H), 2.13 (s, 6H); IR (solid state, cm^{−1}): 1683, 1671, 1593, 1562, 1523, 1476, 1456, 1433, 1398, 1359, 1320, 1289, 1261, 1222, 1183, 1144, 1093, 1070, 1039, 1015, 996, 929, 843, 832, 757, 707, 668, 632, 550; Anal. Calcd. For C₆₂H₄₆Cl₂N₄O₈Zn: C, 67.01; H, 4.17; N, 5.04. Found: C, 66.98; H, 4.17; N, 5.16.

3.6. Determination of the LogP Values

The shake-flask method in combination with UV-Vis spectroscopy was used to determine the LogP values of **2–5**. Briefly, to a suspension of water (500 µL) and octanol (500 µL), **2–5** was added, giving a final concentration of 100 µM. The mixture was shaken for 24 h (room temperature). Using centrifugation, the water and octanol layers were separated and the concentration of **2–5** in each layer was determined using UV-Vis spectroscopy.

3.7. Cell Lines and Cell Culture Conditions

HEK 293T human embryonic kidney cells purchased from American Type Culture Collection (ATCC, Manassas, VA, USA) were grown in DMEM (Dulbecco's Modified Eagle's Medium) with FBS (fetal bovine serum, 10%), and penicillin/streptomycin (1%). Human mammary epithelial CSC-depleted HMLER and human mammary epithelial CSC-enriched HMLER-shEcad cells were obtained from Prof. R.A. Weinberg (Whitehead Institute, MIT). These cells were grown in MEGM (Mammary Epithelial Cell Growth Medium) with BPE, hydrocortisone, hEGF, insulin, and gentamicin/amphotericin-B. All of the cells used in this study were cultured at 37 °C within a humidified incubator with a CO₂ controller to maintain CO₂ levels at 5%.

3.8. MTT (3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) Assay

On day 1, 5000 cells (of the appropriate cell line) were plated in each well of a 96-well plate. The cells were then left to attach to the bottom of the wells overnight. On day 2, the compounds of interest were added (at different concentrations, 0.2–100 µM). The cells were incubated with the compounds for 72 h. The volume of each well at this stage was 200 µL. If DMSO was used to aid solubilization of the compounds, then the same amount of DMSO was also present in control wells. On day 5, a solution of 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) (4 mg/mL) predissolved in PBS was added to each well of the 96-well plate (20 µL). After a further 4 h incubation, the media/MTT solution was removed and the resultant purple formazan crystals were dissolved in DMSO (200 µL). The absorption of each well was measured using a plate reader at 550 nm. The absorption values were normalised according to the control wells and dose–response curves were plotted to determine the IC₅₀ values. Three independent experiments were carried out, with each concentration having six repeats (n = 18).

3.9. Mammosphere Assay

On day 1, 5000 HMLER-shEcad cells were seeded into each well of a 96-well ultralow-attachment plate. The culture media used was MEGM with B27 (Invitrogen), 20 ng/mL EGF, and 4 µg/mL heparin (Sigma). Compounds of interest were added to the HMLER-shEcad cells at various concentrations (0–133 µM), and the suspensions were incubated for 5 days. On day 6, the mammospheres were imaged and counted using a standard microscope. After which, TOX8 (Sigma) (20 µL) was added to each well and incubated for 16 h. On day 7, the solutions in each well were transferred to a black 96-well plate and the fluorescence was measured using a plate reader, 590 nm (λ_{ex} = 560 nm). The arbitrary fluorescence values were normalised with respect to control wells (containing DMSO), and dose–response curves were plotted to determine the IC₅₀ values. Three independent experiments were carried out, with each concentration having three repeats (n = 6).

Supplementary Materials: The following are available online, Figures S1–S9.

Author Contributions: K.S. conceived and designed the experiments; T.K.R., A.E., and S.M.N. performed the experiments; K.S., T.K.R., A.E., and S.M.N. analyzed the data; K.S. and T.K.R. wrote the paper.

Funding: This research received no external funding.

Acknowledgments: T.K.R. was funded through the ERASMUS+ program. A.E. is support by a King's College London Faculty Graduate School International Studentship. S.M.N. was funded through a King's Undergraduate Research Fellowship (KURF). We are grateful to Prof. Robert Weinberg for providing the HMLER and HMLER-shEcad cell lines used in this study.

Conflicts of Interest: The authors declare no conflict of interest.

References

- Cooper, G.M. *The Development and Causes of Cancer*, 2nd ed.; Sinauer Associates: Sunderland, MA, USA, 2000.
- Nguyen, L.V.; Vanner, R.; Dirks, P.; Eaves, C.J. Cancer stem cells: An evolving concept. *Nat. Rev. Cancer* **2012**, *12*, 133–143. [[CrossRef](#)] [[PubMed](#)]
- Marx, J. Cancer's perpetual source? *Science* **2007**, *317*, 1029–1031. [[CrossRef](#)] [[PubMed](#)]
- Pattabiraman, D.R.; Weinberg, R.A. Tackling the cancer stem cells-what challenges do they pose? *Nat. Rev. Drug Discov.* **2014**, *13*, 497–512. [[CrossRef](#)] [[PubMed](#)]
- Plaks, V.; Kong, N.; Werb, Z. The cancer stem cell niche: How essential is the niche in regulating stemness of tumor cells? *Cell Stem Cell* **2015**, *16*, 225–238. [[CrossRef](#)] [[PubMed](#)]
- Yu, Y.; Ramena, G.; Elble, R.C. The role of cancer stem cells in relapse of solid tumors. *Front. Biosci.* **2012**, *4*, 1528–1541. [[CrossRef](#)]
- Dean, M.; Fojo, T.; Bates, S. Tumour stem cells and drug resistance. *Nat. Rev. Cancer* **2005**, *5*, 275–284. [[CrossRef](#)] [[PubMed](#)]
- Abdullah, L.N.; Chow, E.K. Mechanisms of chemoresistance in cancer stem cells. *Clin. Transl. Med.* **2013**, *2*, 3. [[CrossRef](#)] [[PubMed](#)]
- Ishii, H.; Iwatsuki, M.; Ieta, K.; Ohta, D.; Haraguchi, N.; Mimori, K.; Mori, M. Cancer stem cells and chemoradiation resistance. *Cancer Sci.* **2008**, *99*, 1871–1877. [[CrossRef](#)] [[PubMed](#)]
- Kim, J.K.; Jeon, H.Y.; Kim, H. The molecular mechanisms underlying the therapeutic resistance of cancer stem cells. *Arch. Pharm. Res.* **2015**, *38*, 389–401. [[CrossRef](#)] [[PubMed](#)]
- Rich, J.N. Cancer stem cells in radiation resistance. *Cancer Res.* **2007**, *67*, 8980–8984. [[CrossRef](#)] [[PubMed](#)]
- Kaiser, J. The cancer stem cell gamble. *Science* **2015**, *347*, 226–229. [[CrossRef](#)] [[PubMed](#)]
- Laws, K.; Suntharalingam, K. The next generation of anticancer metallopharmaceuticals: Cancer stem cell-active inorganics. *Chembiochem* **2018**. [[CrossRef](#)] [[PubMed](#)]
- Johnstone, T.C.; Suntharalingam, K.; Lippard, S.J. The next generation of platinum drugs: Targeted Pt(II) agents, nanoparticle delivery, and Pt(IV) prodrugs. *Chem. Rev.* **2016**, *116*, 3436–3486. [[CrossRef](#)] [[PubMed](#)]
- Benedetti, M.; Malina, J.; Kasparkova, J.; Brabec, V.; Natile, G. Chiral discrimination in platinum anticancer drugs. *Environ. Health Perspect.* **2002**, *110* (Suppl. 5), 779–782. [[CrossRef](#)] [[PubMed](#)]
- Benedetti, M.; Ducani, C.; Migoni, D.; Antonucci, D.; Vecchio, V.M.; Ciccarese, A.; Romano, A.; Verri, T.; Ciccarella, G.; Fanizzi, F.P. Experimental evidence that a DNA polymerase can incorporate N7-platinated guanines to give platinated DNA. *Angew. Chem. Int. Ed.* **2008**, *47*, 507–510. [[CrossRef](#)] [[PubMed](#)]
- Lunetti, P.; Romano, A.; Carrisi, C.; Antonucci, D.; Verri, T.; De Benedetto, G.E.; Dolce, V.; Fanizzi, F.P.; Benedetti, M.; Capobianco, L. Platinated nucleotides are substrates for the human mitochondrial deoxynucleotide carrier (DNC) and DNA polymerase γ : Relevance for the development of new platinum-based drugs. *ChemistrySelect* **2016**, *1*, 4633–4637. [[CrossRef](#)]
- Suntharalingam, K.; Lin, W.; Johnstone, T.C.; Bruno, P.M.; Zheng, Y.R.; Hemann, M.T.; Lippard, S.J. A breast cancer stem cell-selective, mammospheres-potent osmium(VI) nitrido complex. *J. Am. Chem. Soc.* **2014**, *136*, 14413–14416. [[CrossRef](#)] [[PubMed](#)]
- Tomao, F.; Papa, A.; Rossi, L.; Strudel, M.; Vici, P.; Lo Russo, G.; Tomao, S. Emerging role of cancer stem cells in the biology and treatment of ovarian cancer: Basic knowledge and therapeutic possibilities for an innovative approach. *J. Exp. Clin. Cancer Res.* **2013**, *32*, 48. [[CrossRef](#)] [[PubMed](#)]
- Barr, M.P.; Gray, S.G.; Hoffmann, A.C.; Hilger, R.A.; Thomale, J.; O'Flaherty, J.D.; Fennell, D.A.; Richard, D.; O'Leary, J.J.; O'Byrne, K.J. Generation and characterisation of cisplatin-resistant non-small cell lung cancer cell lines displaying a stem-like signature. *PLoS ONE* **2013**, *8*, e54193. [[CrossRef](#)] [[PubMed](#)]
- Rackham, O.; Nichols, S.J.; Leedman, P.J.; Berners-Price, S.J.; Filipovska, A. A gold(I) phosphine complex selectively induces apoptosis in breast cancer cells: Implications for anticancer therapeutics targeted to mitochondria. *Biochem. Pharmacol.* **2007**, *74*, 992–1002. [[CrossRef](#)] [[PubMed](#)]
- Marzano, C.; Trevisan, A.; Giovagnini, L.; Fregona, D. Synthesis of a new platinum(II) complex: Anticancer activity and nephrotoxicity in vitro. *Toxicol. In Vitro* **2002**, *16*, 413–419. [[CrossRef](#)]
- Gismondi, A.; Nanni, V.; Reina, G.; Orlanducci, S.; Terranova, M.L.; Canini, A. Nanodiamonds coupled with 5,7-dimethoxycoumarin, a plant bioactive metabolite, interfere with the mitotic process in B16F10 cells altering the actin organization. *Int. J. Nanomed.* **2016**, *11*, 557–574. [[CrossRef](#)] [[PubMed](#)]

24. Zhang, Y.; Zhang, L.; Liu, L.; Guo, J.; Wu, D.; Xu, G.; Wang, X.; Jia, D. Anticancer activity, structure, and theoretical calculation of *N*-(1-phenyl-3-methyl-4-propyl-pyrazolone-5)-salicylidene hydrazone and its copper(II) complex. *Inorg. Chim. Acta* **2010**, *363*, 289–293. [[CrossRef](#)]
25. Hu, W.; Fang, L.; Hua, W.; Gou, S. Biotin-Pt (IV)-indomethacin hybrid: A targeting anticancer prodrug providing enhanced cancer cellular uptake and reversing cisplatin resistance. *J. Inorg. Biochem.* **2017**, *175*, 47–57. [[CrossRef](#)] [[PubMed](#)]
26. Boodram, J.N.; McGregor, I.J.; Bruno, P.M.; Cressey, P.B.; Hemann, M.T.; Suntharalingam, K. Breast cancer stem cell potent copper(II)-non-steroidal anti-inflammatory drug complexes. *Angew. Chem. Int. Ed.* **2016**, *55*, 2845–2850. [[CrossRef](#)] [[PubMed](#)]
27. Eskandari, A.; Boodram, J.N.; Cressey, P.B.; Lu, C.; Bruno, P.M.; Hemann, M.T.; Suntharalingam, K. The breast cancer stem cell potency of copper(II) complexes bearing nonsteroidal anti-inflammatory drugs and their encapsulation using polymeric nanoparticles. *Dalton Trans.* **2016**, *45*, 17867–17873. [[CrossRef](#)] [[PubMed](#)]
28. Singh, B.; Berry, J.A.; Shoher, A.; Ramakrishnan, V.; Lucci, A. Cox-2 overexpression increases motility and invasion of breast cancer cells. *Int. J. Oncol.* **2005**, *26*, 1393–1399. [[CrossRef](#)] [[PubMed](#)]
29. Singh, B.; Cook, K.R.; Vincent, L.; Hall, C.S.; Martin, C.; Lucci, A. Role of cox-2 in tumorspheres derived from a breast cancer cell line. *J. Surg. Res.* **2011**, *168*, e39–e49. [[CrossRef](#)] [[PubMed](#)]
30. Sharma, V.; Dixit, D.; Ghosh, S.; Sen, E. Cox-2 regulates the proliferation of glioma stem like cells. *Neurochem. Int.* **2011**, *59*, 567–571. [[CrossRef](#)] [[PubMed](#)]
31. Zhou, Q.; Hambley, T.W.; Kennedy, B.J.; Lay, P.A.; Turner, P.; Warwick, B.; Biffin, J.R.; Regtop, H.L. Syntheses and characterization of anti-inflammatory dinuclear and mononuclear zinc indomethacin complexes. Crystal structures of $[\text{Zn}_2(\text{indomethacin})_4(\text{L})_2]$ ($\text{L} = N,N$ -dimethylacetamide, pyridine, 1-methyl-2-pyrrolidinone) and $[\text{Zn}(\text{indomethacin})_2(\text{L1})_2]$ ($\text{L1} = \text{ethanol, methanol}$). *Inorg. Chem.* **2000**, *39*, 3742–3748. [[PubMed](#)]
32. Deacon, G.B.; Phillips, R.J. Relationships between the carbon-oxygen stretching frequencies of carboxylate complexes and the type of carboxylate coordination. *Coord. Chem. Rev.* **1980**, *33*, 227–250. [[CrossRef](#)]
33. Martinez, D.; Motevalli, M.; Watkinson, M. Is there really a diagnostically useful relationship between the carbon-oxygen stretching frequencies in metal carboxylate complexes and their coordination mode? *Dalton Trans.* **2010**, *39*, 446–455. [[CrossRef](#)] [[PubMed](#)]
34. Jabali, B.; Abu Ali, H. New zinc(II) complexes of the non-steroidal anti-inflammatory drug (indomethacin) and various nitrogen donor ligands. Synthesis, characterization and biological activity. *Polyhedron* **2016**, *117*, 249–258. [[CrossRef](#)]
35. Lakshman, T.R.; Deb, J.; Paine, T.K. Anti-inflammatory activity and enhanced cox-2 selectivity of nitric oxide-donating zinc(II)-NSAID complexes. *Dalton Trans.* **2016**, *45*, 14053–14057. [[CrossRef](#)] [[PubMed](#)]
36. Dontu, G.; Abdallah, W.M.; Foley, J.M.; Jackson, K.W.; Clarke, M.F.; Kawamura, M.J.; Wicha, M.S. In vitro propagation and transcriptional profiling of human mammary stem/progenitor cells. *Genes Dev.* **2003**, *17*, 1253–1270. [[CrossRef](#)] [[PubMed](#)]
37. Lu, C.; Laws, K.; Eskandari, A.; Suntharalingam, K. A reactive oxygen species-generating, cyclooxygenase-2 inhibiting, cancer stem cell-potent tetranuclear copper(II) cluster. *Dalton Trans.* **2017**, *46*, 12785–12789. [[CrossRef](#)] [[PubMed](#)]

Sample Availability: Samples of the compounds 2–5 are available from the authors.



© 2018 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<http://creativecommons.org/licenses/by/4.0/>).